### Water Based Critical Utilities

### **RO, WFI, and Steam**

The bottom of the iceberg that makes a plant run!

**ISPE** 

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## **Start With 4 Simple Questions**

- •What is the starting water quality?
- •What is the water /steam quality that we need?
- •What treatment processes are available and what does each process do?
- •How do I get the water /steam from the point where it is produced to the points where it is used (without picking up contamination along the way) ?



# What public information is available from the local municipality ??

Contaminant Detected	Unit	MCL	MCLG	Level Detected	Range of Detection	Major Sources	Violation
<b>Regulated Contamina</b>	nts						
Nitrate	ppm	10	10	0.34	N/A	Runoff from fertilizer use; leaching from septic tanks, sewage; erosion from natural deposits.	NO
Flouride *(see below)				1.17	0.88 to 1.17	Water additive that promotes strong teeth.	NO
* State (MCL)	ppm	2	none				
* EPA (MCL)	ppm	4	none				
Sodium	ppm	none	none	34.3	N/A	Erosion of natural deposits; road salt, and water treatment chemicals.	NO
Chlorite	ppm	1.0	0.8	0.50	0.21 to 0.50	By-product of drinking water disinfection.	NO
Turbity (see note)	NTU	1.0	TT=100%	0.17	0.06 to 0.17	Soil runoff.	NO

TT= Lowest percentage of monthly samples <0.3 NTU

**Note:** Turbidity is a measure of the cloudiness of the water. We monitor it because it is a good indicator of the effectiveness of our filtration system.

Disinfectant residual	ppm	(MRDL) 4	(MRDLG) 4	.97	0.42 to .97	By-product of drinking water disinfection.	NO
Perchlorate	ppb	2.0	none	0.33	N/A	Rocket propellants, fireworks, munitions, flares, blasting agents. Aged water treatment disinfection chemicals	NO



# What public information is available from the local municipality ??

Contaminant Detected	Unit	MCL	MCLG	Level Detected	Range of Detection	Major Sources	Violation		
Volatile Organic Contaminants									
(TTHM)	ppb	80	0	(50)	0.5 to 50.0	By-product of drinking water chlorination.	NO		
[Total Trihalomethane	al Trihalomethanes] (Highest Runing Annual Average)								
Disinfection By-Produ	ict Co	ontamin	ants						
(HAA)	ppb	60	0	(20.7)	0 to 20.7	By-product of drinking water chlorination.	NO		
[Halo-acetic Acids] (Highest Runing Annual Average)									
Unregulated Contami	nants	5							
MTBE	ppb	none	none	N/D	N/D<0.05	Gasoline Additive.	NO		
Chloroform	ppb	none	none	15.1	3.9 to 15.1	By-product of drinking water chlorination.	NO		
Bromodichloromethane	ppb	none	none	7.3	2.2 to 7.3	By-product of drinking water chlorination.	NO		
Chlorodibromomethane	ppb	none	none	2.5	N/D<0.6 to 2.5	By-product of drinking water chlorination.	NO		
Sulfate	ppm	none	none	5.0	5.0	Mineral and nutrient	NO		



# What public information is available from the local municipality ??

Contaminant Detected	Unit	MCL	MCLG	Level Detected	Range of Detection	Major Sources	Violation
Radionuclides							
Gross Alpha	pCi/l	15	0	0.5 (+-1.1)	N/A	Erosion of natural deposits	NO
Radium 228	pCi/l	5	0	0.1 (+-0.6)	N/A	Erosion of natural deposits	NO
Contaminant	Unit	MCL	MCLG	Level Detected	Range of Detection	Major Sources	Violation
Lead	ppb	15	0	.001	0 of 50	Corrosion of household plumbing systems. Erosion of natural deposits.	NO
Copper	ppm	1.3	1.3	0.04	0 of 50	Corrosion of household plumbing systems. Erosion of natural deposits; Leaching from wood preservatives.	NO

Finished water pH ranged from 7.5 to 8.3



## What we really need to know

Parameter	Method	Result	MRL
- City Water			
Sampled: 10/2/2013 by Clier	nt		
Barium, MG/L	EPA 200.7	0.024	0.001
Boron, MG/L	EPA 200.7	0.02	0.01
Calcium, MG/L	EPA 200.7	38.2	1
Magnesium, MG/L	EPA 200.7	2.8	1
Potassium, MG/L	EPA 200.7	4.2	1
Silica as SiO2, MG/L	EPA 200.7	8.5	1
Sodium, MG/L	EPA 200.7	99.3	1
Strontium, MG/L	EPA 200.7	0.181	0.001
Ammonia, MG/L	SM 4500-NH3-D	ND	0.1
Bicarbonate, MG/L	SM 2320B	ND	1
Carbonate, MG/L	SM 2320B	75	1
Chloride, MG/L	EPA 300.0	164	1
Fluoride, MG/L	EPA 300.0	ND	0.1
Nitrate as N, MG/L	EPA 300.0	0.24	0.05
Sulfate, MG/L	EPA 300.0	10.8	1

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## Let's understand what has to be removed Classify the various contaminants

- Particles or Suspended Solids
- Dissolved Solids
  - Ionized
  - Non-ionized
  - •Organic
- Colloidal Materials
- Dissolved Gases
- Bacteria and other living organisms

## **Contaminants introduce variability !!**



### **Particles or Suspended Solids**

Materials that do not dissolve in water

Can be any shape

Mostly considered as hard, spherical particles

Moving water holds more particles

Larger and more dense particles will settle out by themselves

Smaller particles may never settle



### **Dissolved solids, lonized**

Materials that dissolve in water

Form free floating ions in solution

Adds positive and negative charges to a solution

Solution remains electrically neutral

The ionized solids content changes how much electricity the water can conduct

Direct relationship between the abundance of ions and the conductivity of the water



### **Dissolved solids, Non-Ionized**

Materials that dissolve in water

Do not form free floating ions in solution

- No charge is added to the solution
- No change in the conductivity of the solution
- Cannot measure abundance by measuring conductivity



## **Dissolved solids, Organic**

Materials that dissolve in water and contain carbon

May add a slight charge to a solution

Small in size, so these materials are very difficult to remove from a solution

Cannot measure abundance by measuring conductivity



## **Colloidal Materials or Suspensions**

Contain carbon

- Large in molecular size (10,000-5,000,000 MW) Slightly negative charge
- Somewhere between suspended and dissolved

Too small to settle by themselves Held in solution by size and charge repulsion Undetectable change in the conductivity Measure abundance by silt density index Can plug up purification processes



### **Dissolved Gases**

#### Nitrogen, oxygen, carbon dioxide, etc.

Not removed by most purification processes

More dissolved gases in solution at lower temperatures (opposite of dissolved solids)

Least understood and least studied contaminant

Carbon dioxide is troublesome because it ionizes when it dissolves into solution

Ammonia can be troublesome to some purification processes in waters treated with chloramine

Present as a contaminant in clean steam as non condensible gases



### **Bacteria and other living organisms**

#### Not uniformly distributed in a water system

- Exist in equilibrium with their environment
- More food = more bacteria
- Less than 1% is free floating (detectable)
- Vast majority is present as biofilm
- Compete for Nutrients with the cells we want



Where does our water come from? How do it properties vary?

## Well Water Surface Water

Low Suspended Solids High Dissolved Salts Low Colloidal Content Some Dissolved Gases High Suspended Solids Low Dissolved Salts High Colloidal Content High Dissolved Gases



## What water quality do we really need ? It depends !

Where are we in the product's life cycle ?

## Drug Discovery

## Research Pilot Scale

## **Clinical Trials**

## Full Scale Manufacturing



#### Labs use CLSI/NCCLS or ASTM specifications for purity

			ASTM			Μ	
TYPE 1	TYPE 2	TYPE 3	TYPE 1	TYPE 2	TYPE 3	TYPE 4	
<0.1	<0.2	<0.5	0.056	1.0	0.25	5.0	
>10.0	>2.0	>1.0	18.0	1.0	4.0	0.2	
						5.8-8.0	
<500	<100	<1000	3	3	500		
			1	5	10	50	
			1	5	10	50	
			100	50	200		
<10	10		Separate specification, only where bacteria control is required Type 1 : 10/1,000 ml Type 2 : 100/1,000 ml Type 3 : 10,000/1,000 ml				
	TYPE 1 <0.1 >10.0  <500  	TYPE 1       TYPE 2         <0.1	TYPE 1       TYPE 2       TYPE 3         <0.1	TYPE 1       TYPE 2       TYPE 3       TYPE 1         <0.1	TYPE 1         TYPE 2         TYPE 3         TYPE 1         TYPE 2           <0.1	TYPE 1         TYPE 2         TYPE 3         TYPE 1         TYPE 2         TYPE 3           <0.1	

#### **Dialysis has their own requirements**

#### CHEMICAL CONTAMINANTS & MAXIMUM ALLOWED (MG/L

Aluminum	0.01	Lead	0.005
Antimony	0.006	Magnesium	
Arsenic	0.005	Mercury	0.0002
Barium	0.10	Nitrate	2.0
	0.0004	Potassium	
Beryllium			8 (0.2 mEq/L)
Cadmium	0.001	Selenium	0.09
Calcium	2 (0.1mEQ/L)	Silver	0.005
Chloramines	0.10	Sodium	70 (3.0 mEq/L)
Chromium	0.014	Sulfate	100.0
Copper	0.10	Thallium	0.002
Fluoride	0.20	Zinc	0.10
Free Chlorine	0.50		
BACTERIA		MAXIMUM ALLO	WED
Water used for di	alysate →	<200 CFU/ml	
(RD52,4.1.2)	-	Endotoxin level <	2 EU/ml
Dialysate $\rightarrow$	$\rightarrow$ $\rightarrow$	<200CFU/ml	
(RD52, 4.3.2.1)		Endotoxin level <	2 EU/ml



#### **Microelectronics requirements are unbelievable !**

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PARAMETER	ATTAINABLE	ACCEPTABLE	ALERT	CRITICAL
Resistivity	18.2	18.2	17.9	17.5
TOC (online, ppb)	<1	<2	5	10
THM (ppb)	<2	<5		
Particles by laser 0.05 to 0.1 micron 0.1 to 0.2 micron 0.2-0.3 micron 0.3-0.5 micron >0.5 micron	<100/1000 ml <50/1000 ml <20/1000 ml <10/1000 ml <1/1000 ml	<500/1000 ml <300/1000 ml <50/1000 ml <20/1000 ml <4/1000 ml		
Bacteria (cfu/1000 ml)	<1	<6	25	>25
Silica (total, ppb)	<0.5	<3	>5	>10



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PARAMETER	ATTAINABLE	ACCEPTABLE	ALERT	CRITICAL
Phosphate (ppb)	<0.02	<0.1	>0.01	>0.5
Silicate (ppb)	<0.05	0.1	<0.02	>0.5
Sodium (ppb)	<0.01	0.05	>0.02	>0.5
Potassium (ppb)	<0.02	<0.1	>0.02	>0.5
Ammonium (ppb)	<0.06	0.1	<0.02	>0.5
Calcium (ppb)	<0.02	<0.1	>0.01	>0.2
Magnesium (ppb)	<0.02	<0.1	<0.01	>0.2
Fluoride (ppb)	<0.1	<0.1	>0.02	>0.5
Chloride (ppb)	<0.02	0.1	<0.02	>0.5
Bromide (ppb)	<0.02	<0.1	>0.01	>0.5
Nitrate (ppb)	<0.02	<0.1	<0.01	>0.5



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METAL ION (	CONTAMINANTS,	ALL ARE MEASU	RED IN PARTS PI	ER TRILLION
Aluminum (ppt)*	7	50	>0.0	200
Barium (ppt)*	2	10	>50	100
Boron (ppt)*	300	<2000		
Chromium (ppt)*	8	30	>30	50
Copper (ppt)*	5	50	>50	>200
Iron (ppt)*	10	100	200	>200
Lithium (ppt)*	4	30	100	>100
Magnesium (ppt)*	2	20	100	>200
Manganese (ppt)*	4	30	>30	100
Nickel (ppt)*	5	50	>50	100
Sodium (ppt)*	10	60	>200	>500
Strontium (ppt)*	2	10	>10	>10
Zinc (ppt)*	8	60	>50	>100



## **Pharmaceutical Water Quality**

PARAMETER	USP PURIFIED	USP WFI
Total Organic Carbon (ppb)	500	500
Conductivity	<1.3 @ 25°C	<1.3@25°C
Bacteria	None given, but recommended to be 100/ml	None given, but recommended to be 10/100 ml
Endotoxins		<0.25 EU/ml

## Hey, Why Is Injectable Grade Water Allowed To Have Bacteria ??



#3 What water purification processes are available?

## What does each one actually DO?

# Particle filters remove contaminants based on their size

Ion exchange removes contaminants based on their charge



Carbon filters remove small (below 1,000 MW) non polar molecules

#### **Remove disinfectants from drinking water**

Protects chlorine sensitive reverse osmosis membranes



### Ultraviolet units come in two basic flavors

Single wavelength units (254 nm) for bacterial control

Dual wavelength units (185 & 254 nm) for organics and bacteria control

Dual wavelength units (185 & 254 nm) increase the conductivity of the water, so location is extremely important



### Ozone Generators are becoming more popular

Oxidizes organics Kills bacteria Consumes biofilm

But, misapplication and misuse of ozone technology has led to problems, making many users reluctant



Distillation is the only water treatment process that removes the water from the contaminants

Considered the gold standard for producing Water-For-Injection (WFI) grade water

Dissolved gases and some chemicals can carry over into distillate (product water)



## Summary of Unit Operations

	Coarse Particle Filters	Absolute Membrane Filters	Ultrafilters	Reverse Osmosis	Carbon Filtration	Ultraviolet Disinfection	Deionization
Particles	F	G-E	E	Е	N	Ν	N
Dissolved lons	N	N	N	G-E	N	N	Е
Small Organics	Ν	N	Ν	F-G	G-E	Ν	Р
Colloids	Ν	F-P	G-E	Е	P-F	Ν	Р
Bacteria	Р	Е	Е	Е	A	G	A-P



## Sequencing of Unit Processes Varies between equipment manufacturers

Remove Particles first Suspended Solids Colloidal materials

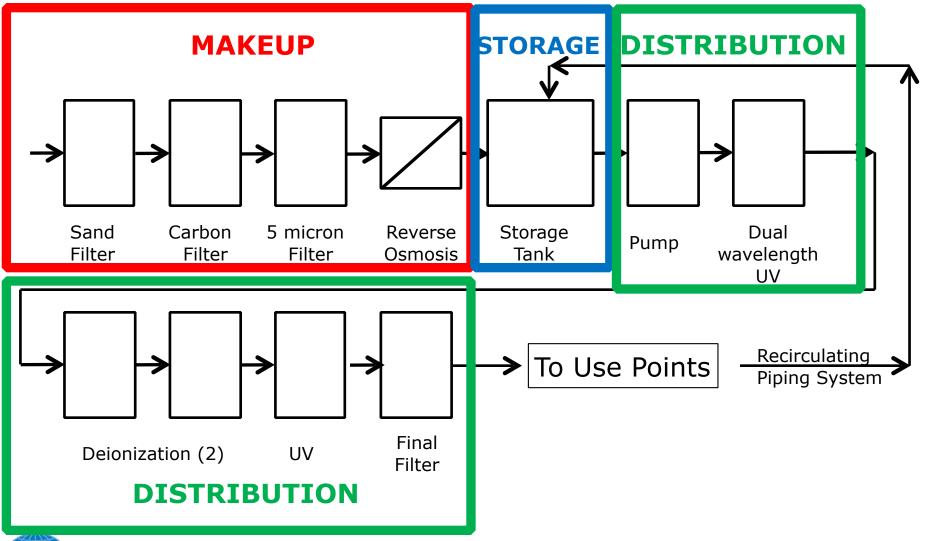
Remove dissolved ions next

Remove trace materials Ions, organics, particles System generated impurities

Remove bacteria as the last step



## Sequencing of Unit Processes Varies between equipment manufacturers





## Design of Distribution Piping Systems

Design around 5 feet per second (FPS) velocity Design for 3 FPS in return with use points active No dead legs (6D rule) WFI water almost always piped in stainless steel Purified water can be piped in SS, PP, PVDF





## **Steam Systems**



## **Two Types of Steam Systems**

### Plant or Utility Steam

### Clean or Pure Steam



## **Plant or Utility Steam**

Produced by a Carbon Steel Utility Boiler Contains Additives to Prevent Corrosion Elevated pH of ~10 Pressures in excess of 60 psi are common



## **Clean or Pure Steam**

Produced by a Clean Steam Generator Characterized by No Additives pH below 7 15 psi pressure Used in Pharmaceutical or Biotech Plants for Thermal Disinfection

(10<sup>5</sup> reduction)

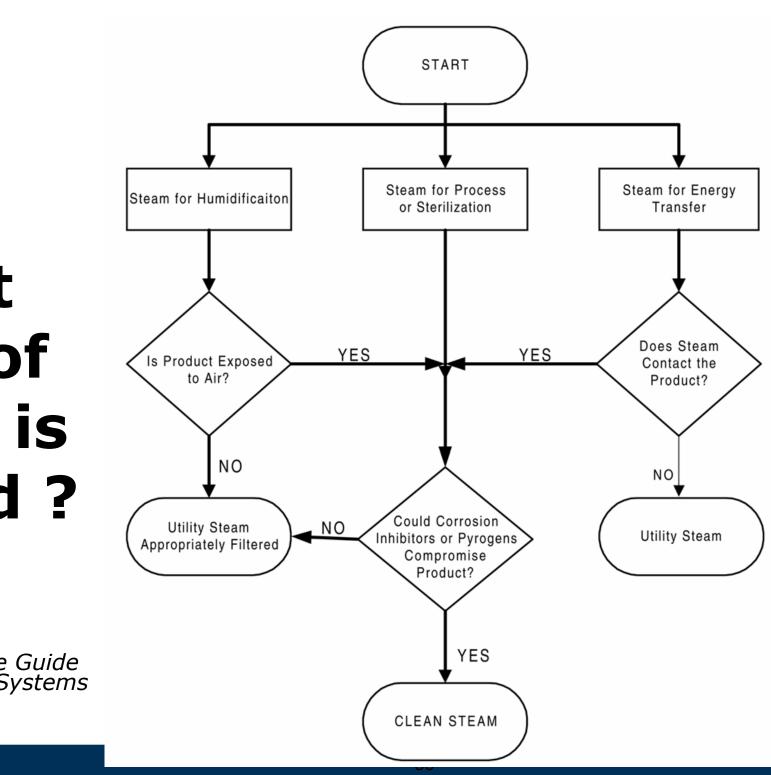
Sterilization (10<sup>6</sup> reduction)



## What Type of Steam is Needed ?

*Source: ISPE Baseline Guide for Water and Steam Systems* 

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## Contaminants of Concern Clean or Pure Steam

Non Condensable Gases (3.5% Max) Creates cool spots Water Droplets (Dryness 0.90 Min) Entrained moisture or condensate from high velocities Superheated steam (25°C max over sat.)

Purity specified in British std EN 285

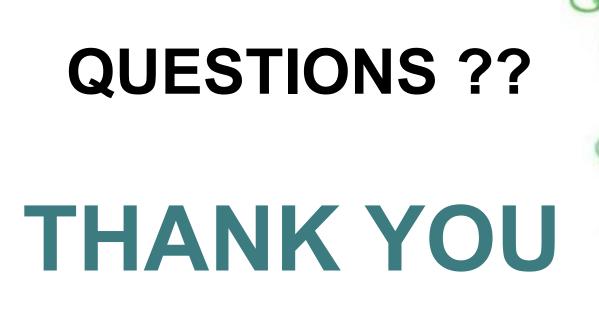


## Design of Distribution Piping Systems for Clean or Pure Steam

Surface finish is not as critical here Slope in direction of flow Maximum design velocity of 120 feet/sec Ball valves are common practice Condensate traps before use points Non condensable gas vent at high point or in vessel







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