

Protein A Resin Lifetime Study

Evaluation of Protein A Resin Performance with a Model based approach in continuous capture

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Biopharmaceuticals







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Protein A resin



- Found in cell membrane of S.aureus
- Functionally made up of 5 domains



- Affinity binding resin Capture step
- Various recombinant forms available
- Provides > 95% product purity



Disadvantage: Expensive & Protein A degradation



Protein A resin in a packed column

Your Drug



Source: Applied biosystems & GE healthcare

- The column is packed with resin particle
- Each resin particle has numerous pores
- Within each pore are the Protein A ligands
- The ligands are bound to the resin at multiple locations via linker
- Antibody in the feed specifically binds to the Protein A ligand attached to the column



Impact of caustic on Protein A

NaOH – popular in biopharmaceuticals Breaks down proteins and saponifies fats Inactivates bacteria, yeasts, endotoxins, etc. Cost-effective



Cause of Protein A degradation







Protein A resin lifetime - Problems

- How do the Protein A ligands degrade?
- What is the effect of NaOH on the Protein A ligands?
- How to compensate for the ligand loss in column switching decision?

Poor understanding of the column after multiple uses



Overview: Protein A resin lifetime study





Degradation reaction

- leaching of Protein A ligands
- unfolded or denatured Protein A ligands still bound to the resin particle and
- unfolded or denatured Protein A ligands removed during sanitization and elution phase of a typical column run.

Modified Shrinking Core Model (MSCM)



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- Packed column comprises of porous resin particles.
- Each particle has tortuous pore channels
- Protein A ligands bound to the stationary phase matrix.

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Model Equations

Fraction of degraded Protein A is given by
Reaction rate at interface
Number of degradable Protein A particles consumed
Reaction rate
Number of degradable Protein A particles available in the resin is given by

 $R = 1 - \frac{r^3}{r_0^3}$ $D\frac{dC}{dr} = \alpha k C^n$

$$\frac{dN}{dt} = -4 \pi r^2 C^n k$$

 $r_e = -kC^n$

 $\frac{dN}{dt} = 4\pi r^2 \frac{dr}{dt}$

R = Fraction of degraded Protein A r = radius of unreacted core region (cm) r_0 = initial radius of resin particle (cm) C = NaOH concentration at a given time (M) D = diffusion constant (cm²/s) k = Rate constant K = k/D α = stoichiometry factor N = number of degradable Protein A particles reacted in time t n = reaction order

The model equations were numerically solved using the ode23 solver in MATLAB.

K, n and D parameters estimated using MATLAB

Source for Shrinking Core model: Advanced Separation textbook , Hsu et al 1975



Lifecycle study – experimental protocol





Breakthrough Run Analysis

Various curves represent incubation time of resin in caustic: 0hr (black), 15hr (blue), 30hr (red), 45hr (yellow) and 60hr (green)

The breakthrough curve at higher loading provides a clear distinction of the loading pattern

Variables:

- UV measurement -- Antibody concentration
- Conductivity
- pH
- Pressure



Behere et al, Manuscript under review, Biotech Progress



Retained Binding Capacity

All the experiments were performed in duplicates for reproducibility.

The column performance was assessed at 70%, 80% and 90% of binding capacity.

Binding capacity steadily decreased from 100% to 45%



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Model Validation

Resin	Reaction	Rate	Diffusion						
	order	constant	constant (cm²/s)						
Resin A	3	4E-08	5E-08						
Resin B	4	4E-08	5E-08						
Resin C	1.65	4E-08	5E-08						

The experimental data at 70% of retained binding capacity was used to perform the model validation and reaction order estimation



The reaction order depends on the complex reaction mechanism occurring within the resin pores

Behere et al, Manuscript under review, Biotech Progress



Model Simulations

Predict the behavior of the binding capacity

Different concentration of caustic (0.1N, 0.2N, 0.3N, 0.4N and 0.5N) used

Cycle number	0.1M NaOH	0.5M NaOH						
120	R	R						
Resin A	0.01	0.33						
Resin B	0.01	0.32						
Resin C	0.03	0.62						



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Conclusion

Reaction order

- key indicator to qualify the resin performance
- Assess proposed cleaning strategies

Cause of resin degradation

- linker-ligand breakdown
- ligand conformation changes

Significance of higher order reactions

- increased OH⁻ ions are required to perform hydrolysis
- Reduced degradation rate





Industry Applications

Process Development

- Evaluate resin stability to caustic
- Make column switching decision
- Assess excipient type and concentration to improve the Protein A ligand stability
- Soft sensor for online column performance monitoring Regression isotherm toolbox
- Batch and continuous operations
- Ability to make real-time decision for column performance



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